The Canadian High Arctic: a Window into Climate, Ozone & Global Air Quality

Dan Weaver

Sigma Xi Distinguished Lecture Series 19 November 2015

Introductions





Dan Weaver

- Ph.D. Candidate, Dept. of Physics
- Travel to PEARL for annual winter/spring campaigns

(since 2012)

• I also run campaign communications Motivation for atmospheric research in the high Arctic?

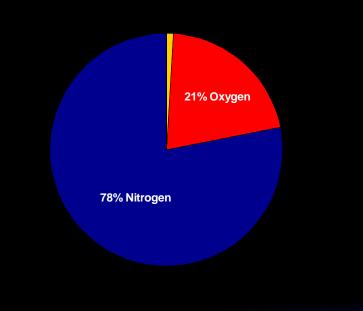
Understanding planetary atmospheric and environmental change



Why study the atmosphere?

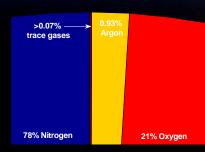
Image taken from the International Space Station, above Alaskan Aleutian Islands

What is air?



Trace Gases include:

- Argon
- Water vapour
- Carbon dioxide
- Methane
- Nitrous oxide
- Ozone
- Many others



Small amounts of gases can have a BIG impact on the planet!

Why is this research important now?







Painting by Elizabeth Simcoe, from the City of Toronto "*The History of Toronto: An 11,000-Year Journey*" web page



2 December 2

Downtown Toronto

1890s

2014

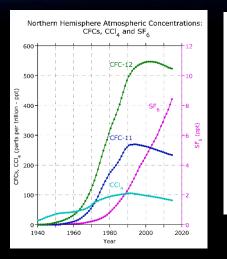
Earth at night, 2015

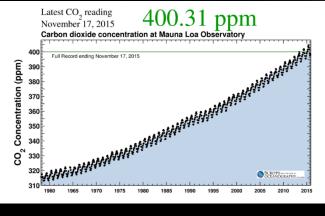


Credit: NASA

Atmospheric science

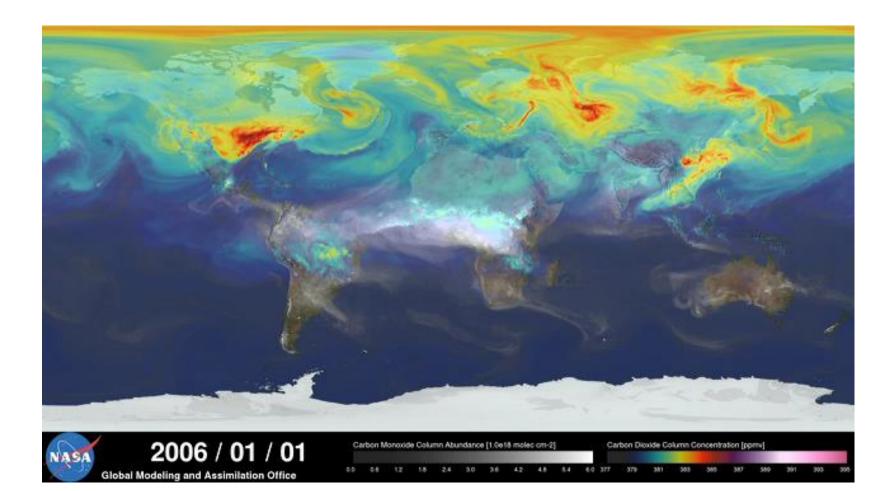
aims to measure & understand the changing atmosphere





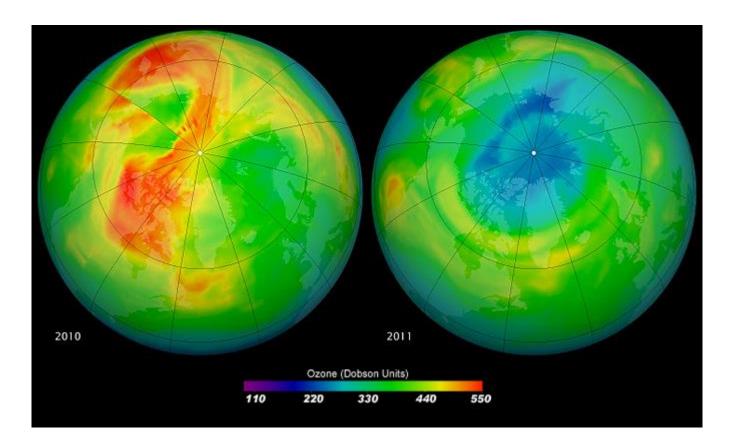


CO_2

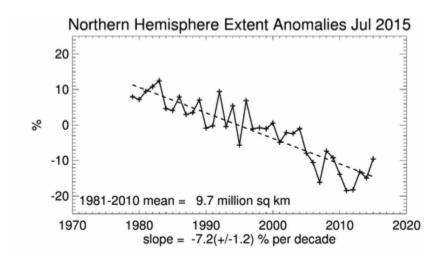




Ozone depletion



Arctic sea ice change



Dwindling Arctic Sea Ice

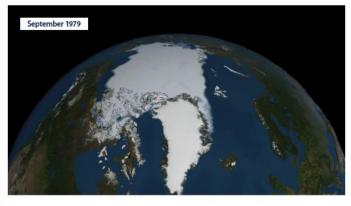
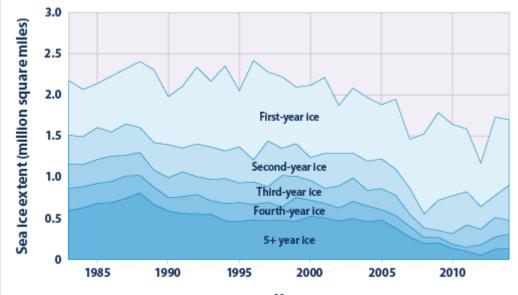




Figure 2. Age of Arctic Sea Ice at Minimum September Week, 1983–2014



Year

Our society's impressive rapid technological development has come at a significant environmental cost



we should at least understand how we are changing the planet

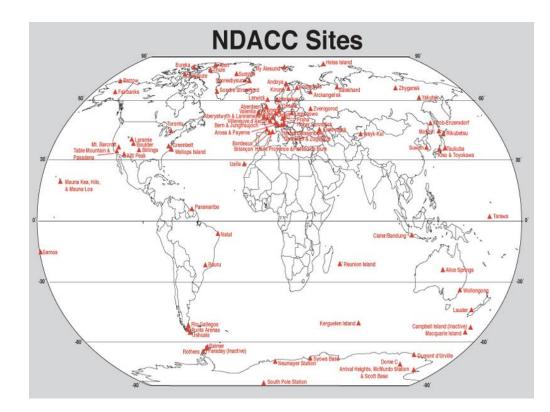
Ideally, we should avoid them entirely

Today's talk

Where is Eureka PEARL research Working in the High Arctic

Global context





There are atmospheric observatories world-wide

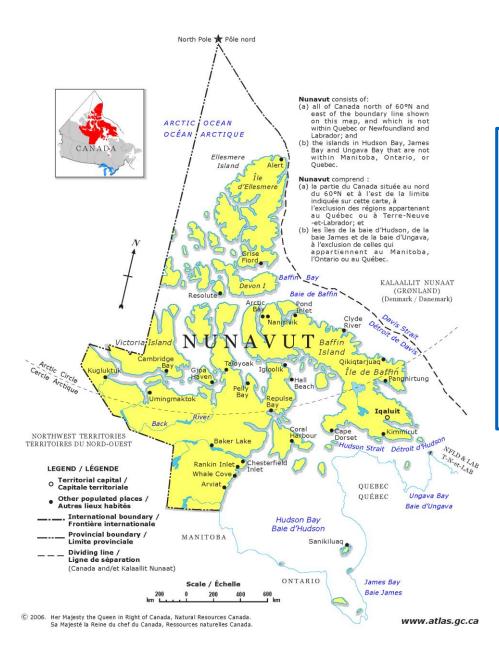
... but there are gaps!



Our Arctic is 40% of Canada's landmass & is home to over 100,000 Canadians.

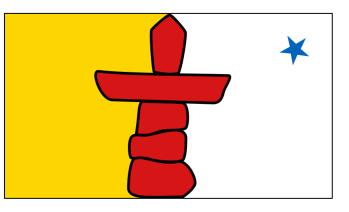
Canadian Arctic

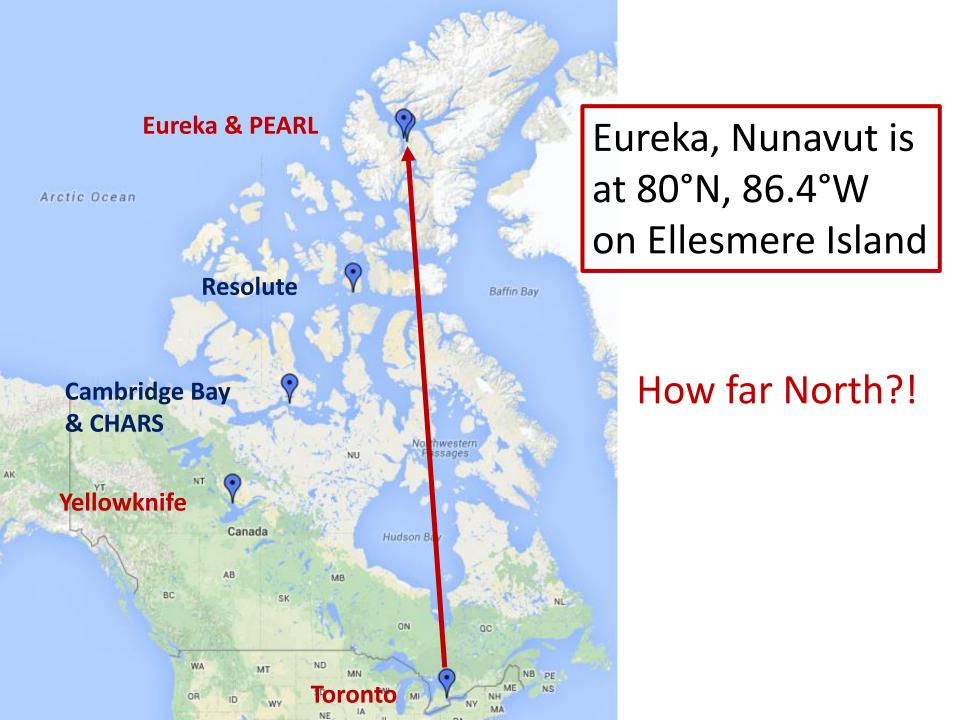




Nunavut

- Territory created in 1999
- Over 80% Inuit population (total: ~37,000)
- Smallest economy, largest area of all provinces/territories
- Elects a single MP
- Has distinct government system (consensus)
- Unique culture (Inuit)







Eureka & PEARL

Arctic Ocean

Baffin Bay

~4000 km

thwesten

QC.

Hudson B

Cambridge Bay ?

Yellowknife

BC

WA

Canada

AB



MT ND MN SD Toronto Eureka, Nunavut is at 80°N on Ellesmere Island

Distances

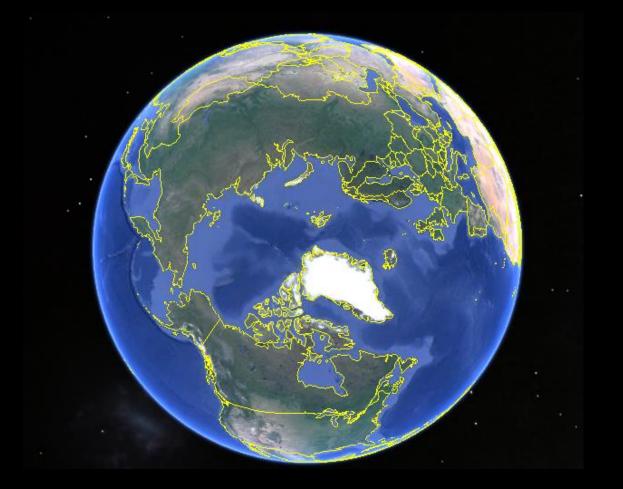
Eureka/PEARL is over 4000 km north of Toronto!

Yellowknife is ~halfway point

.....

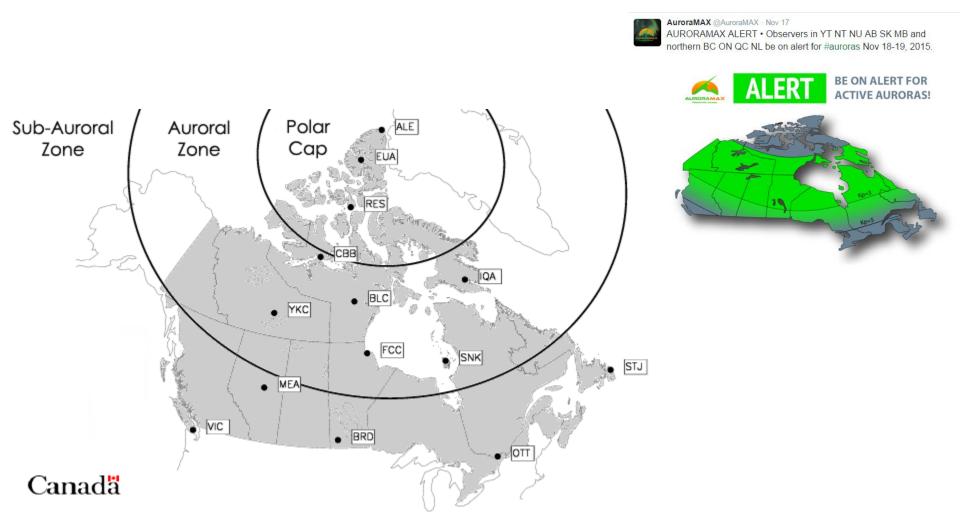
Eureka is closer to:

Stockholm, Sweden St. Petersburg, Russia Reykjavik, Iceland Glasgow, Scotland

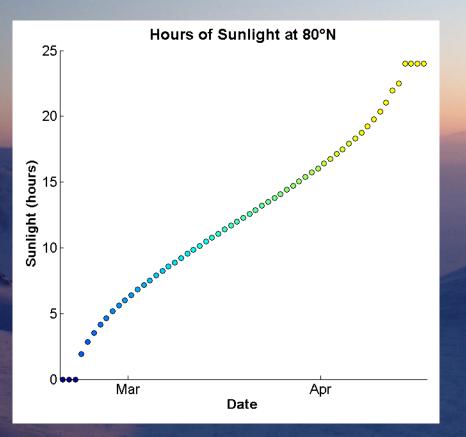


Than to Toronto!

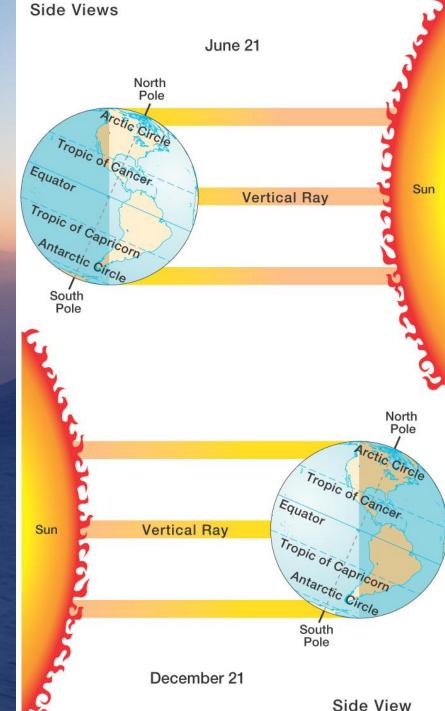
Eureka is farther north than most Northern Lights!

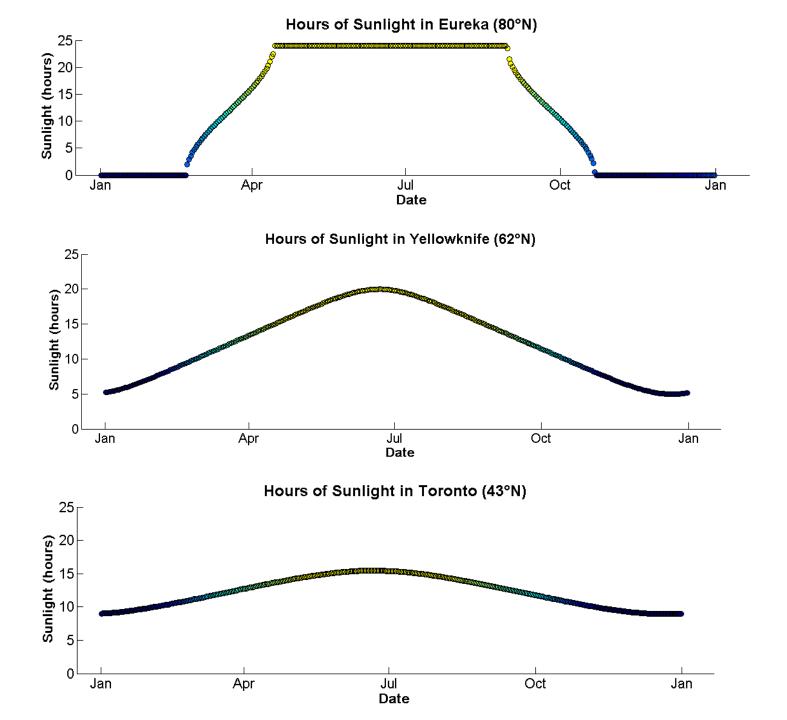


Daylight



Eureka, Nunavut 18 February – 18 April 2015 Data from the U.S. Naval Observatory





Travelling to PEARL...

Toronto Edmonton Yellowknife Cambridge Bay Resolute Eureka







Eureka International Airport





Dress warmly!



NORMAL

MIN

MA

TEMPERATURE / MAX. TEMP.

DEWPOINT/MIN. TEMP.

A.E.S. REMOTE TEMPERATURE / DEWPOINT DISPLAY

ERROR MESSAGE POWER

CLEAR

Eureka

- Environment Canada facility
- 8 permanent staff
- Short-term research teams



Eureka at twilight

Hydrogen shed

Eureka Weather Station Diesel & power generation Zero-altitude PEARL auxiliary lab (OPAL)

The Eureka Weather Station







Eureka is the northernmost permanently settled civilian place in the world*



*Nord, Greenland and Alert, Nunavut are further north, but are primarily military It's like a tiny town. There is:

- Post office
- Store
- Bar (BYOB)
- Airport
- Labs
- Guest rooms
- Power facilities
- Sewage facilities
- Water infrastructure
- Hospital/clinic

Eureka Weather Station's recreation room



There's always plenty of food in Eureka

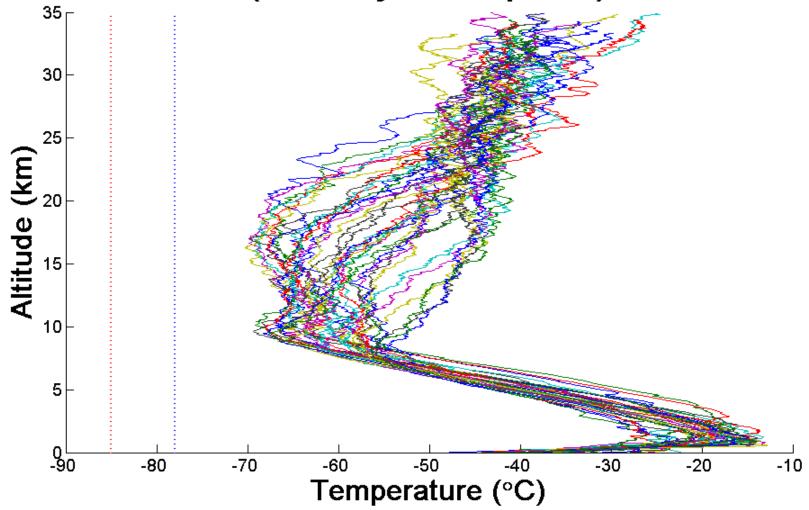




Diesel power generators & temperature inversions

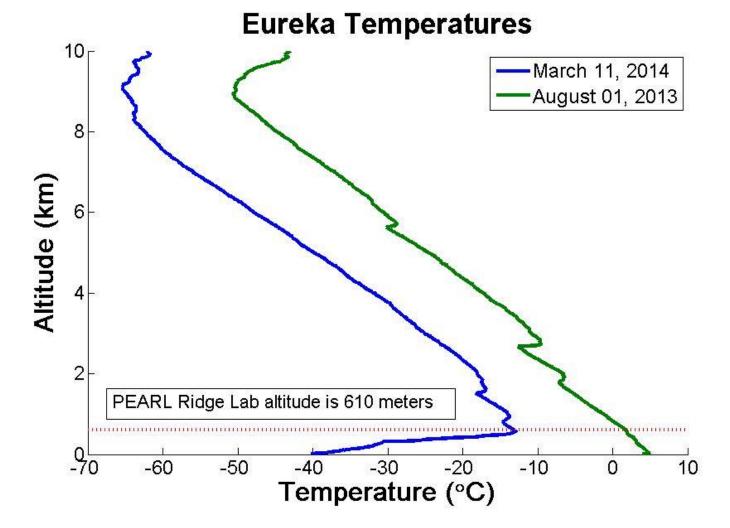


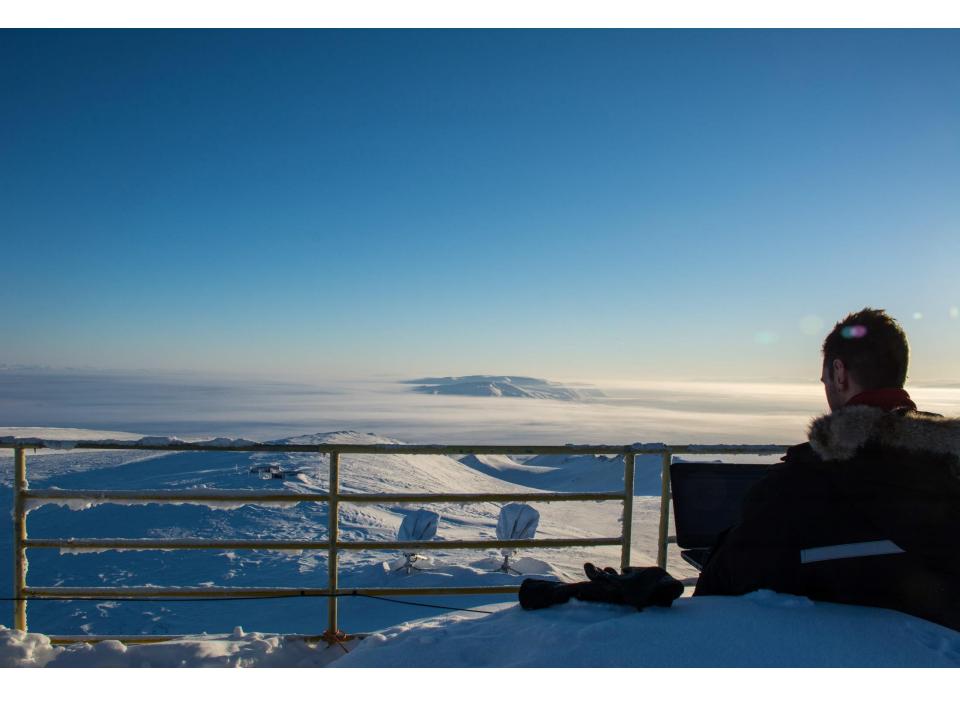
2014 ACE Campaign radiosonde profiles (February 26 to April 02)





Interesting phenomena: Temperature inversion











Driving to the PEARL Ridge Lab

Be prepared to shovel snow when in the Arctic!

DYK:

Canada's high Arctic is a polar desert. There is very low humidity!



Figure .: Location of PEARL relative to Eureka.



EC has a variety of specialized vehicles...





Along the road to the PEARL Ridge Lab



PEARL



- Ideally situated to study the high Arctic atmosphere
- Run by CANDAC since 2006
- Originally built as an Environment Canada facility, AStrO, in the 1990s

(Arctic Stratospheric Ozone Observatory)



- 80.05° North, 86.4° West
- 610 meters a.s.l.
- ~15 km from Environment Canada's Eureka Weather Station

The PEARL Ridge Lab





It's not easy to get here, but it's beautiful!

Zero-altitude PEARL Auxiliary Lab (0PAL)







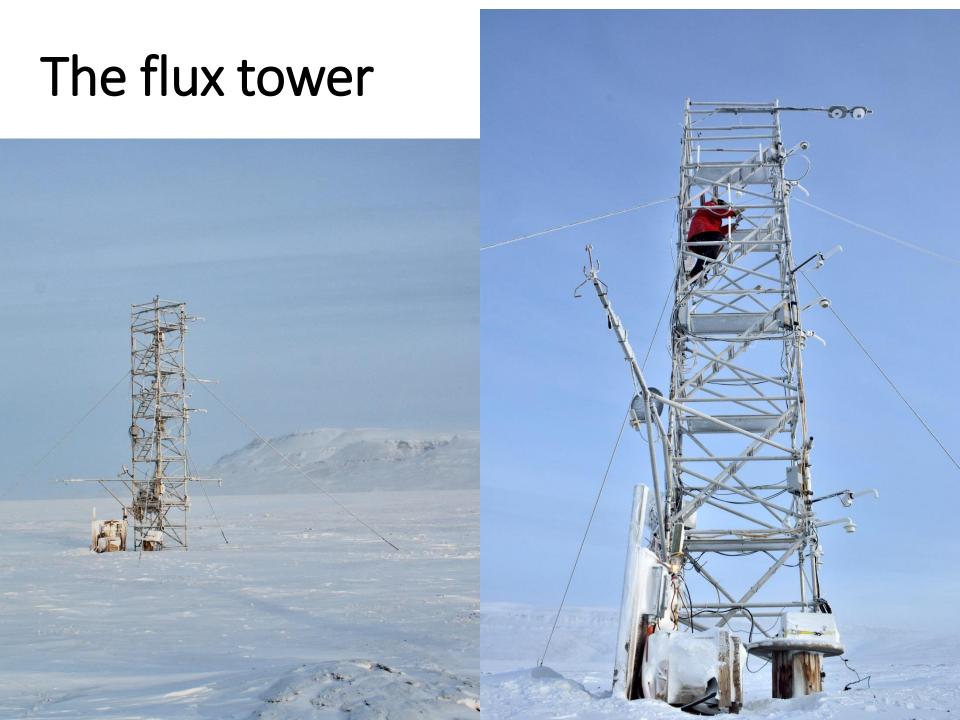




SAFIRE

(Surface and Atmospheric Flux, Irradiance and Radiation Extension)





Research at PEARL?

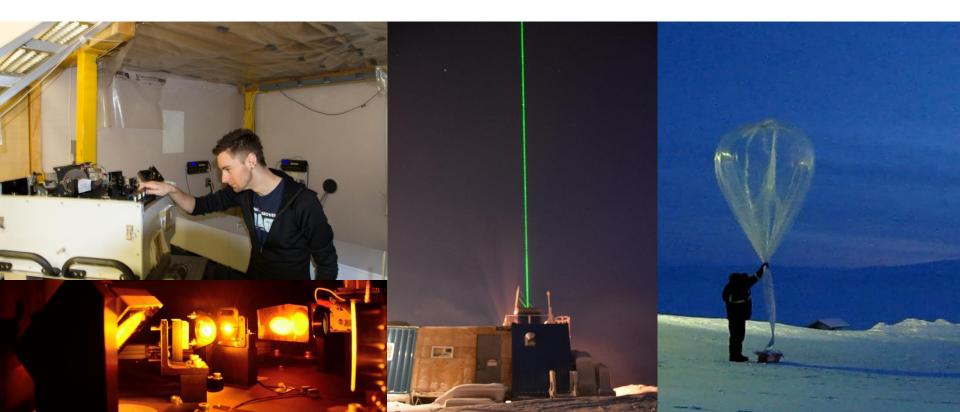
Ozone Depletion Climate Change Pollution

How do we measure the atmosphere?

Ground-based spectrometers

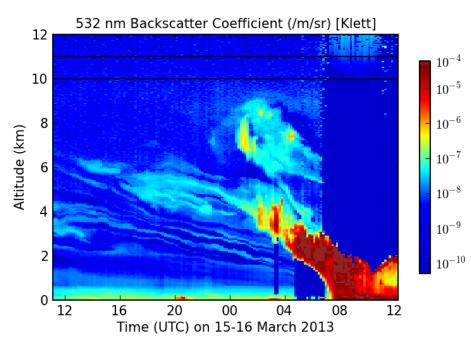
Ground-based lidars

Balloon measurements



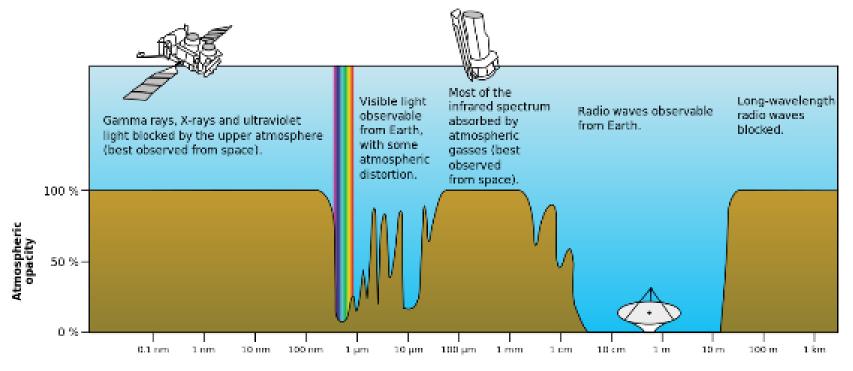
LIDARs

Light detection and ranging





PEARL's instruments mostly observe the atmosphere in the UV, visible, infrared and microwave regions of the EM spectrum



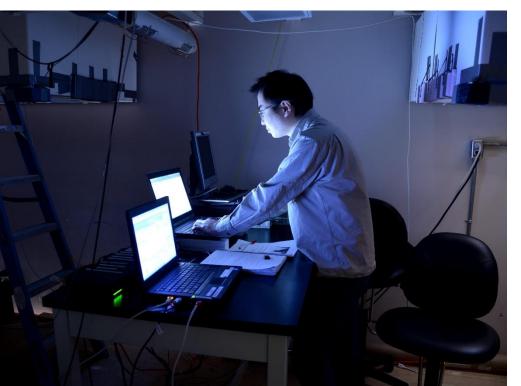
Wavelength

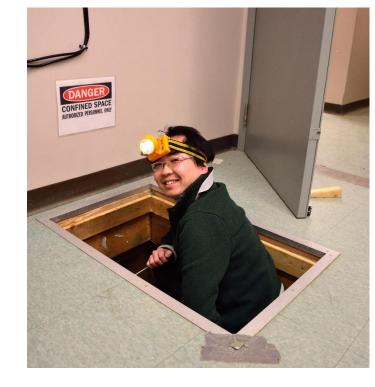
E-AERI: an emission FTIR





UV-Vis Lab









PEARL 125HR

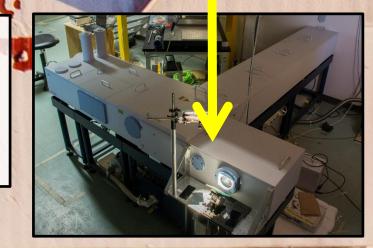
Uses sunlight to create "maps" of the atmosphere



PEARL 125HR

Solar absorption spectrometer

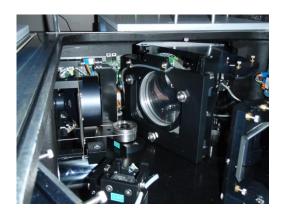
- MIR and NIR measurements
- KBr & CaF₂ beamsplitters
- NDACC & TCCON instrument
- Solar tracker built in-house

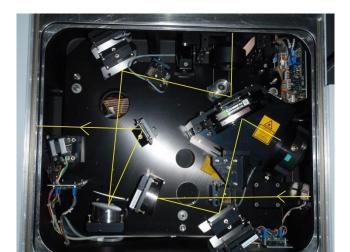


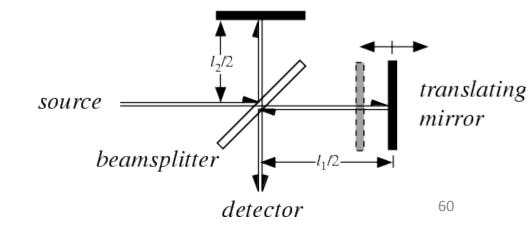
* Sun is not to scale

Fourier transform spectroscopy

- Light is passed through a beam splitter
- Two beams of light travel different path lengths
- Detector measures an interference pattern
- As the moving mirror changes position, different frequencies of light peak at different times. The result of a scan is an interferogram.
- The Fourier transform of the interferogram is the observed spectrum, which can be analyzed to retrieve gas abundances and distribution.
- The spectral resolution of an FTS is directly related to the maximum optical path difference







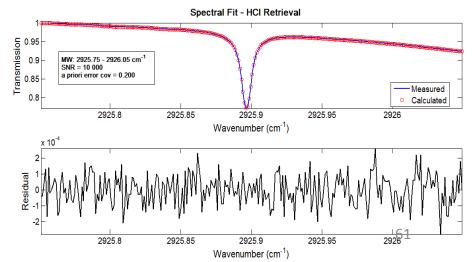
fixed mirror

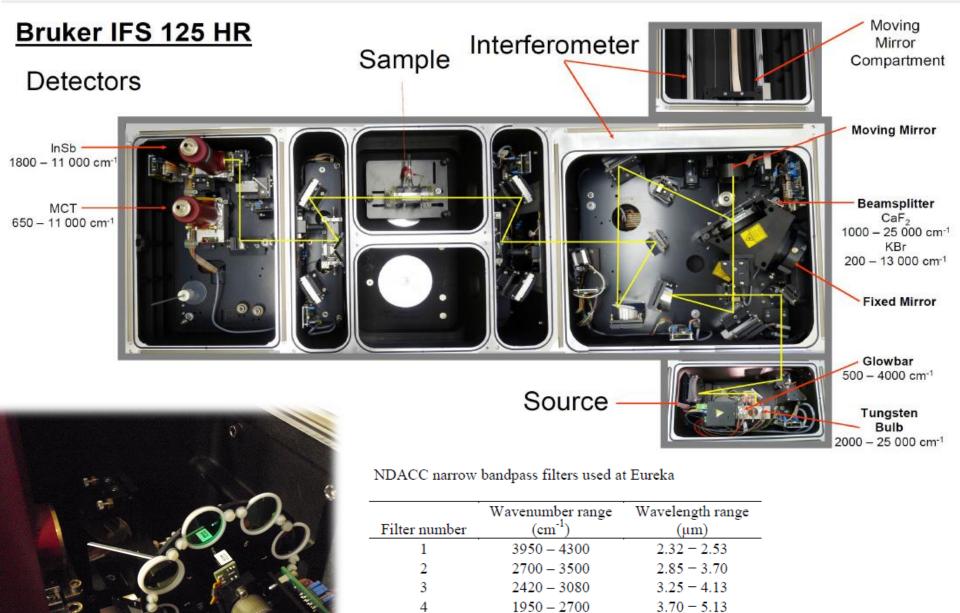
Spectral features

- Extensive lab tests provide a database of spectral features for atmospheric gases.
 - There are 64 000 water vapour lines!
- This knowledge enables us to select appropriate features to examine.
- Inferring abundances and vertical distribution relies on the shape of the observed spectral line.
 - The line strength is directly related to the abundance.
 - The shape is related to the distribution.

$$S = \int_{-\infty}^{\infty} k_{\nu} \, d\nu$$

where k_{ν} is the absorption coefficient of the molecule being studied





* these filters were installed in August 2007.

1800 - 2200

700 - 1350

700 - 1000

1000 - 1400

4.55 - 5.55

7.40 - 14.28

10.00 - 14.28

7.14 - 10.00

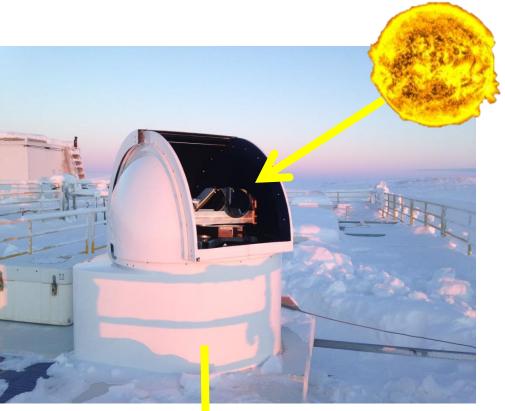
5

6

7*

8*

Table from Lindenmaier (2012)

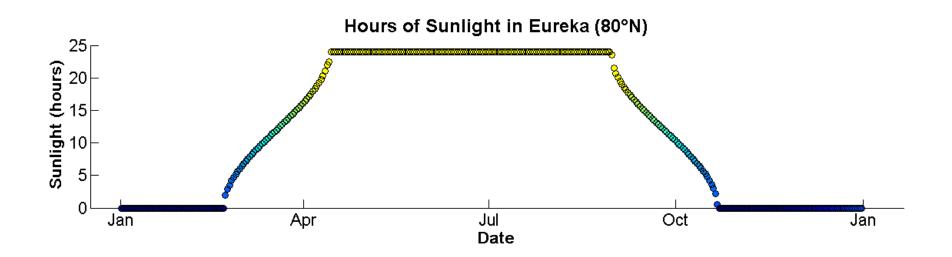








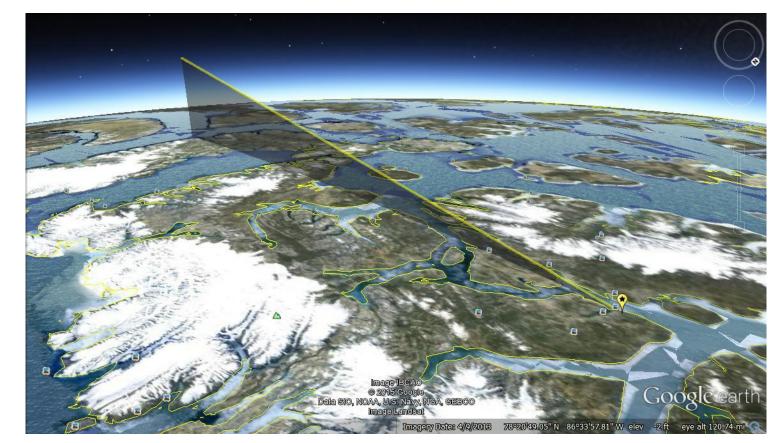
While a powerful technique, FTIRs are limited to clear weather, sunny conditions.



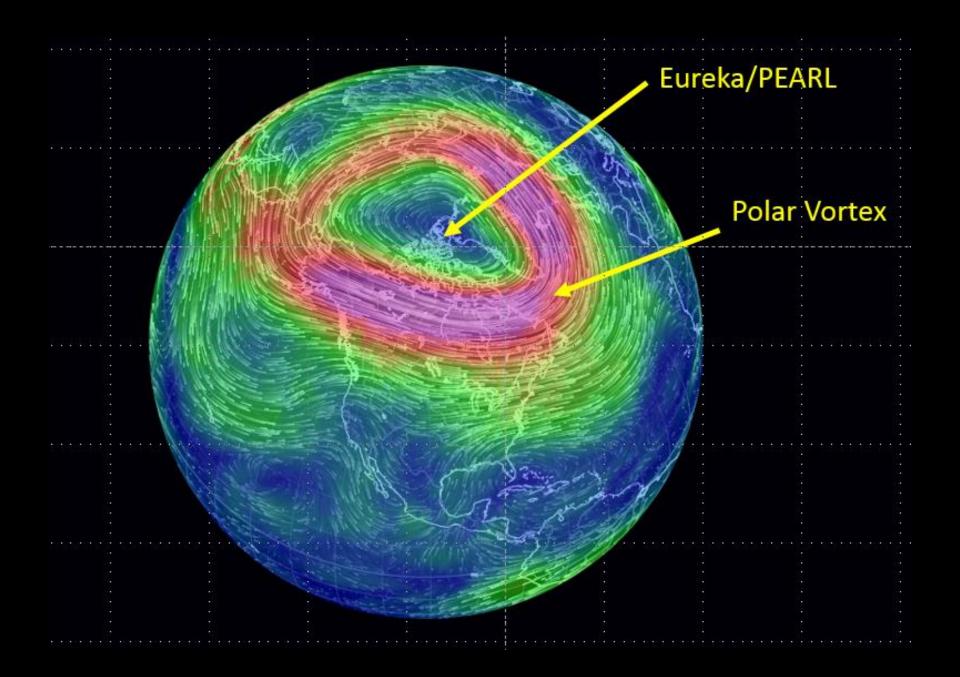
We are currently experimenting with using the moon as a light source...



Because the FTIR is looking at the Sun from a high latitude, we're actually looking south

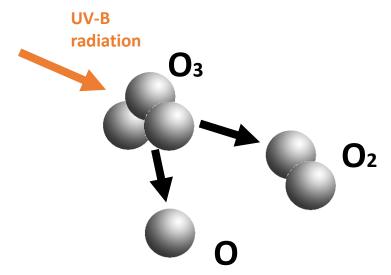


At 8 km altitude, we're measuring ~ 70km away At 20 km altitude, we're measuring ~150 km away At 35 km altitude, we're measuring ~250 km away At 50 km altitude, we're measuring ~350 km away At 100 km altitude, we're measuring ~600 km away

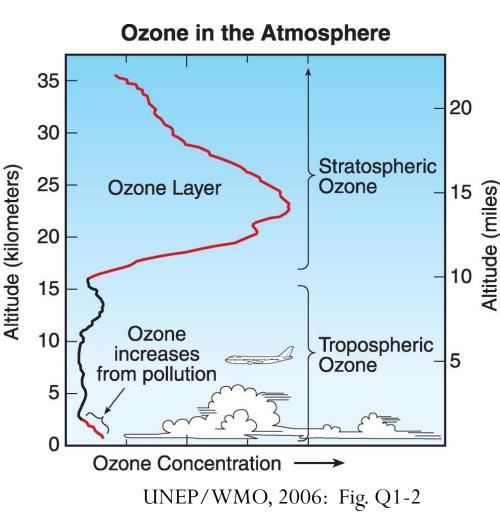


What are we using these measurements to study?

Ozone



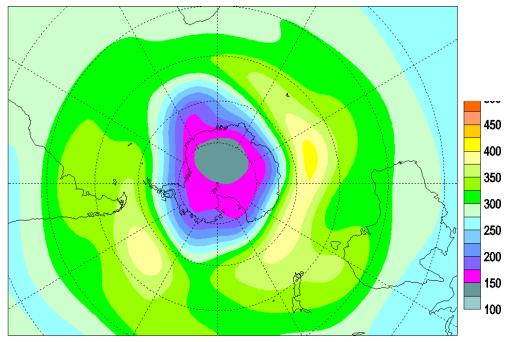
- Ozone in the stratosphere protects us from harmful UV radiation (good!)
- Ozone in the troposphere is a pollutant and harmful to breath! (bad!)



"Space isn't remote at all. It's only an hour's drive away if your car could go straight upwards." – Sir Frederick Hoyle

Ozone 'Hole' in Antarctic: First Discovered in 1985

Mean total ozone (DU), 2010/10/01-2010/10/31



What is an Ozone 'Hole'?

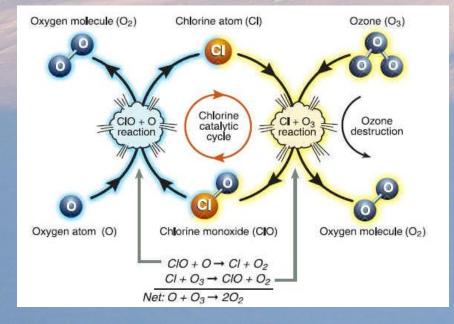
- When there is very *little* ozone
- ~ 1/3 (or less) than usual
- Puts plants and animals at greater risk of UV radiation exposure.
- Too much UV radiation can damage skin cells, which can lead to skin cancer and strong UV radiation can damage the cell structure of plant material.

http://exp-studies.tor.ec.gc.ca/cgi-

bin/selectMap?lang=e&type1=du&day1=mm&month1=10&year1=2010&howmany1=1&interval1=1&intervalunit1=m&hem1=s&type2=no&day 2=23&month2=02&year2=2010&howmany2=1&interval2=1&intervalunit2=d&hem2=n&mapsize=100

How CFCs destroy ozone

 When CFCs reach the ozone layer, ultraviolet sunlight breaks them apart to form atomic chlorine.
 CF₂Cl₂ + sunlight ----> CF₂Cl + Cl



One chlorine atom can destroy <u>thousands</u> of ozone molecules, because it is "recycled" through catalytic cycles.

Ending Ozone Destruction

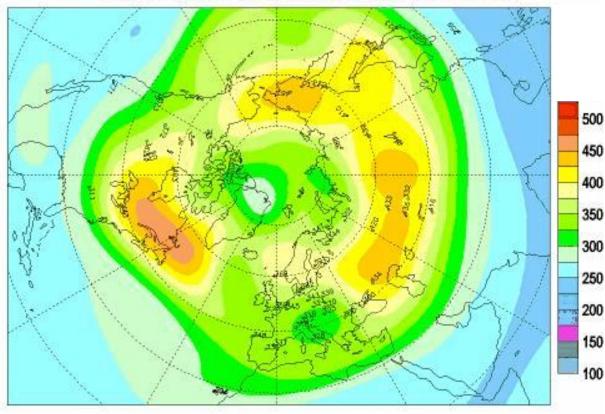
How does this stop?

• When CI encounters methane, CH₄ or when CIO encounters NO₂, the cycle terminates

$$CI + CH_4 ----> HCI + CH_3$$
$$CIO + NO_2 ----> CINO_3$$

A Big Surprise: 2011 Arctic Ozone Hole

Total ozone (DU) / Ozone total (UD), 2011/02/10



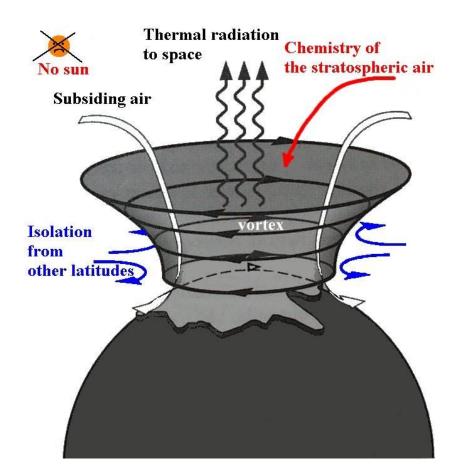


Colder than usual temperatures in the Arctic stratosphere caused the largest ozone depletion event yet observed. Debate surrounds use of the term 'hole'.

Why Ozone Holes in the polar regions?

The Polar Vortex enables the stratosphere to reach very low temperatures (< 78°C), allowing Polar Stratospheric Clouds to form.

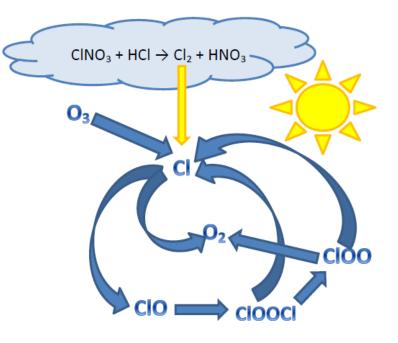
PSCs provide a surface for ozone-destroying chemical reactions to happen.

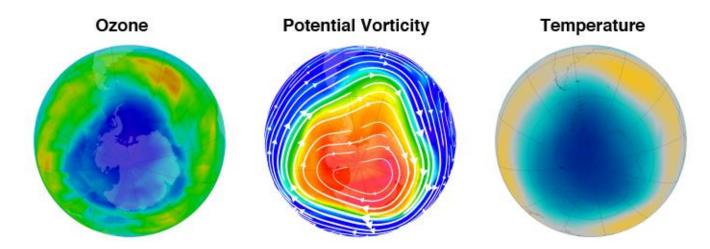


PSC chemistry

 $CIONO_{2} + HCI \rightarrow CI_{2} + HNO_{3}$ $CI_{2} + sunlight \rightarrow 2 CI$ $CI + O_{3} \rightarrow O_{2} + CIO$

Without PSCs, $CIONO_2$ and HCl would not generate active chlorine, but instead would remain in a reservoir until eventually deposited to the ground.

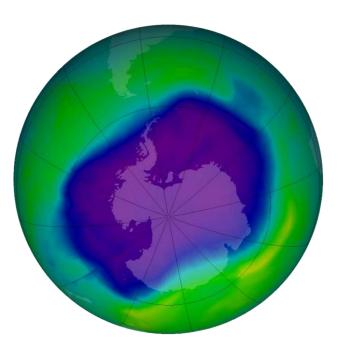




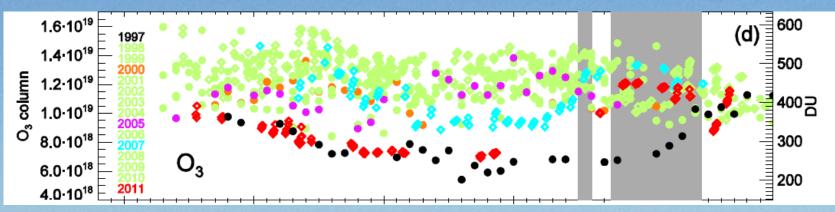
Why is ozone depletion *still* a problem?

- CFCs have a long lifetime.
- It will take at least another 50 years for a full recovery
- This should be a warning for us to be cautious about modifying the Earth's atmosphere...

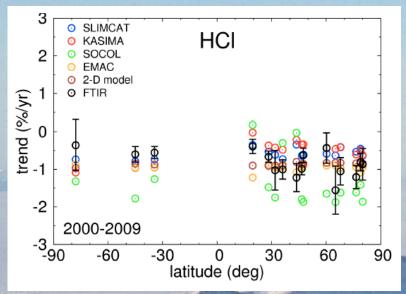


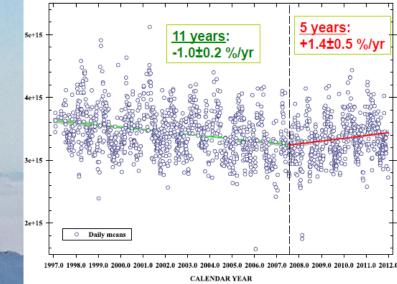


Current research into trends



PEARL O₃ timeseries from Lindenmaier (2012)





Global HCl trends from Kohlhepp (2012)

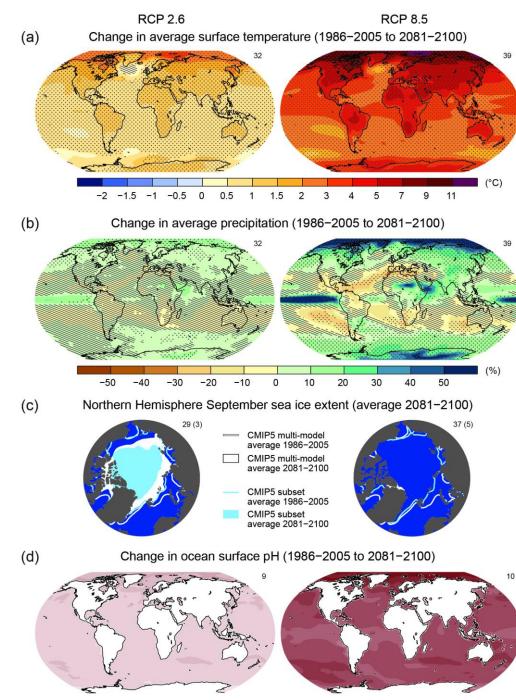
Jungfraujoch HCl from Mahieu (2012)

FTIRs have a useful advantage

- They can take large spectra from which *many* atmospheric gases can be measured simultaneously
- This is a powerful means of observing chemistry.
- PEARL's FTIR measurements are being used to investigate conversions between active and reservoir forms of Chlorine, and the relationships with nitrogen chemistry

Climate change

- Over the last 50 years, the Arctic has warmed by 5°C (Lesins et al. 2010)
- The Arctic is expected to warm more than & faster than any other area of the world. (IPCC, 2014)



^{-0.6 -0.55 -0.5 -0.4 -0.35 -0.3 -0.25 -0.2 -0.15 -0.1 -0.05}

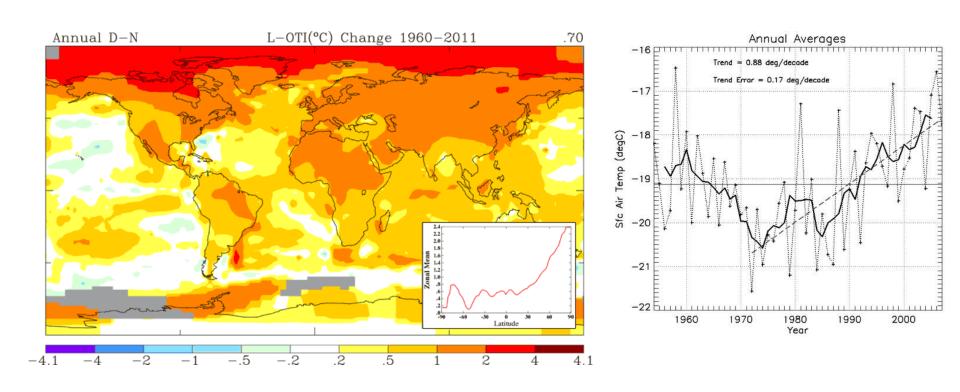
Arctic temperature change

Eureka temperature trend: 0.88°C/decade since 1972

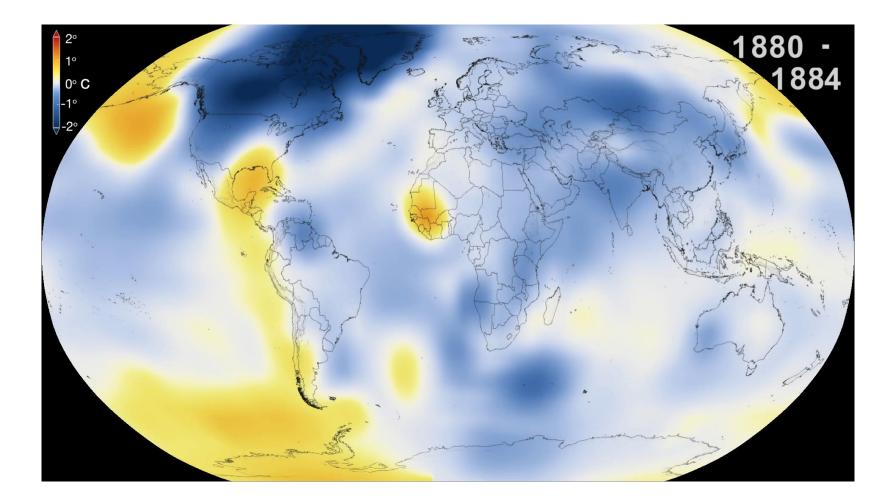
(Lesins et al. 2010)

Total warming:

- 1.5°C summer (least)
- 4.5°C autumn (most)



Global temperature change



Total Carbon Column Observing Network (TCCON)



Arctic climate & water vapour

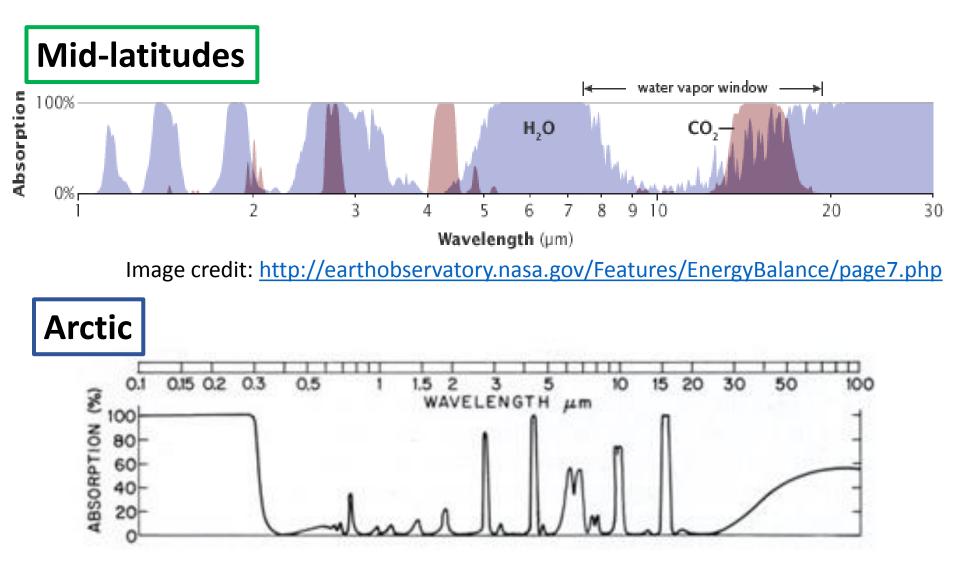
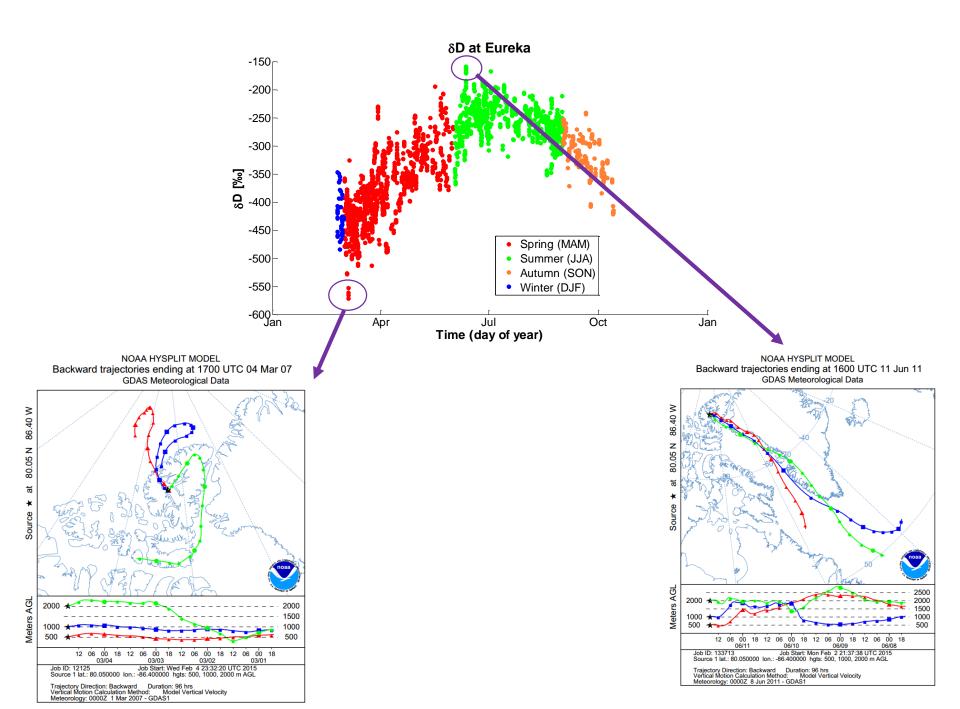
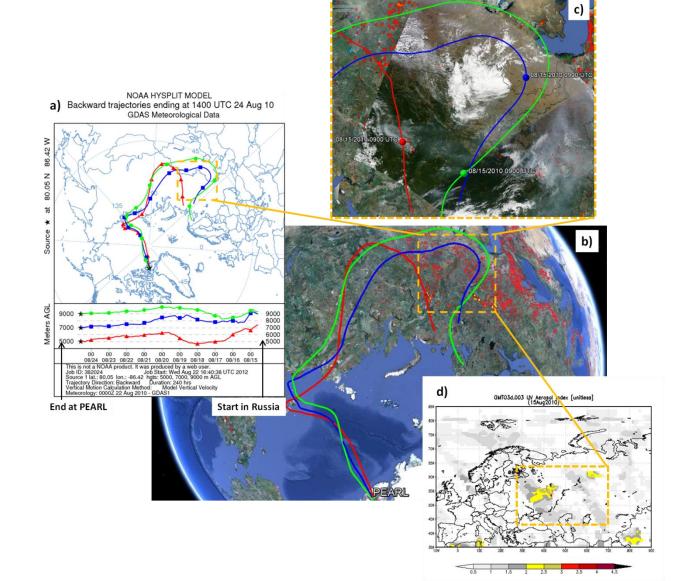


Image credit: Zak, B. and K. Stamnes (1997): ARM Science Meeting – Session Papers. 19-25



Pollution transport



One more science objective...

Satellite validation

Eureka is easy to get to ...for a satellite!

Sun-synchronous orbits provide global coverage, and frequently pass over the poles

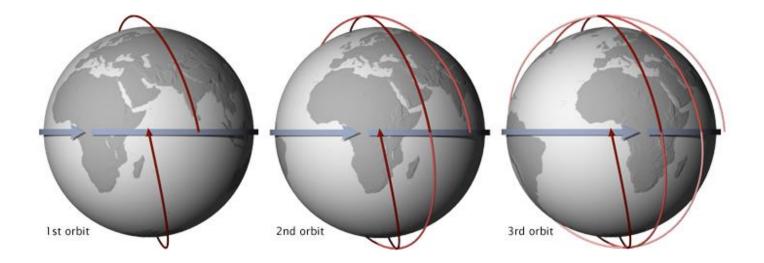


Figure from http://earthobservatory.nasa.gov/Features/OrbitsCatalog/page2.php

ACE (Canada)



OCO-2 (U.S. NASA)



Odin (Sweden & Canada)

GOSAT (Japan)







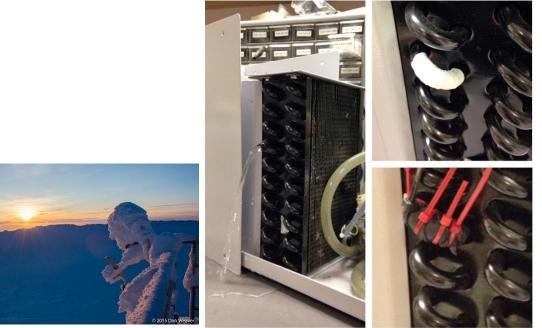
PARIS

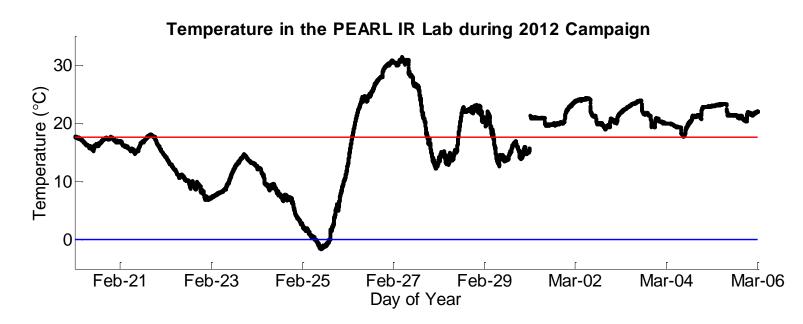


Working in the extreme cold is challenging, for people & equipment



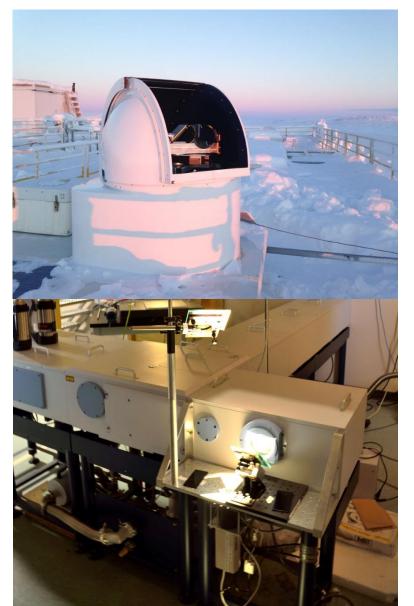
Operational challenges abound







Upgrades

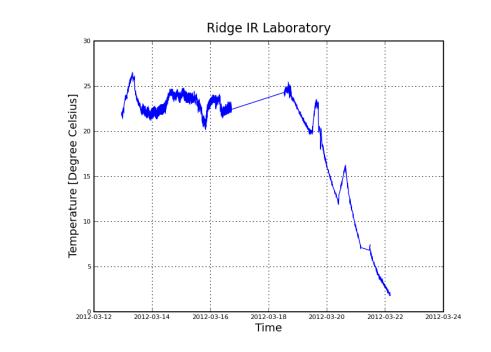




Heat & Cooling







Arctic challenges



Internet

- Website for CANDAC, which runs PEARL
- Campaign website
- Social media
 - Twitter
 - Instagram
 - blog





Twitter: jokes with NASA engineer & former internet meme



Bobak Ferdowsi @tweetsoutloud · Feb 27 Is this the rebel base on Hoth? MT"@csa_asc: @createarcticsci: 1st day at PEARL pic.twitter.com/NSPfP0iyde"

🗊 Hide photo

🔦 Reply 😘 Retweeted 🜟 Favorited 🚥 More 📲 HootSuite



RETWEETS	FAVORITES
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& more fun



Thank you.



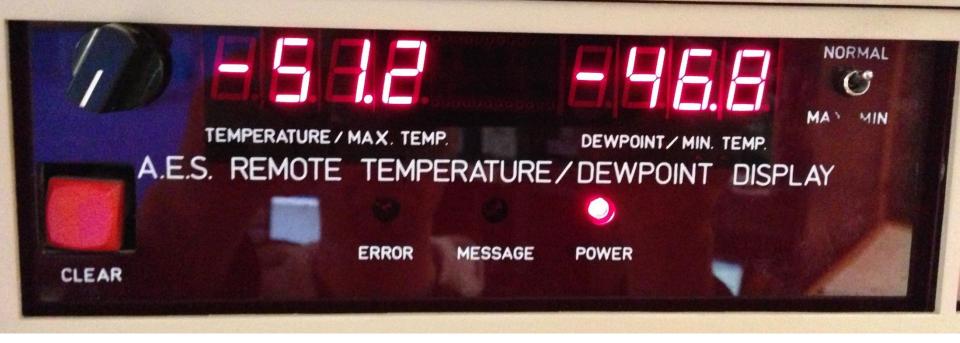
More information <u>www.twitter.com/CREATEarcticsci</u> <u>www.instagram.com/CREATEarcticsci</u> <u>http://createarcticscience.wordpress.com/</u> <u>http://acebox.uwaterloo.ca/eureka/</u> <u>CANDAC.ca</u>

Contact me <u>dweaver@atmosp.physics.utoronto.ca</u>

Exploring Eureka



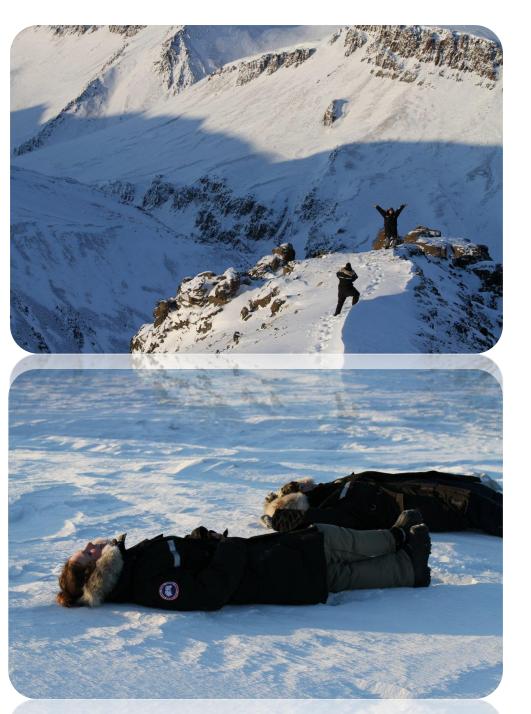
Temperature

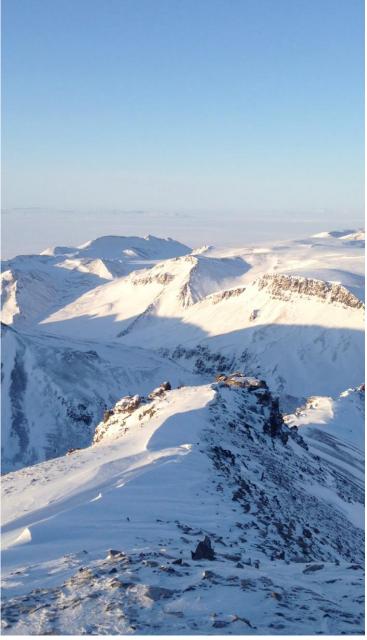


Dress warmly!

Bring a sense of adventure & curiosity











Night hikes





Arctic Wildlife





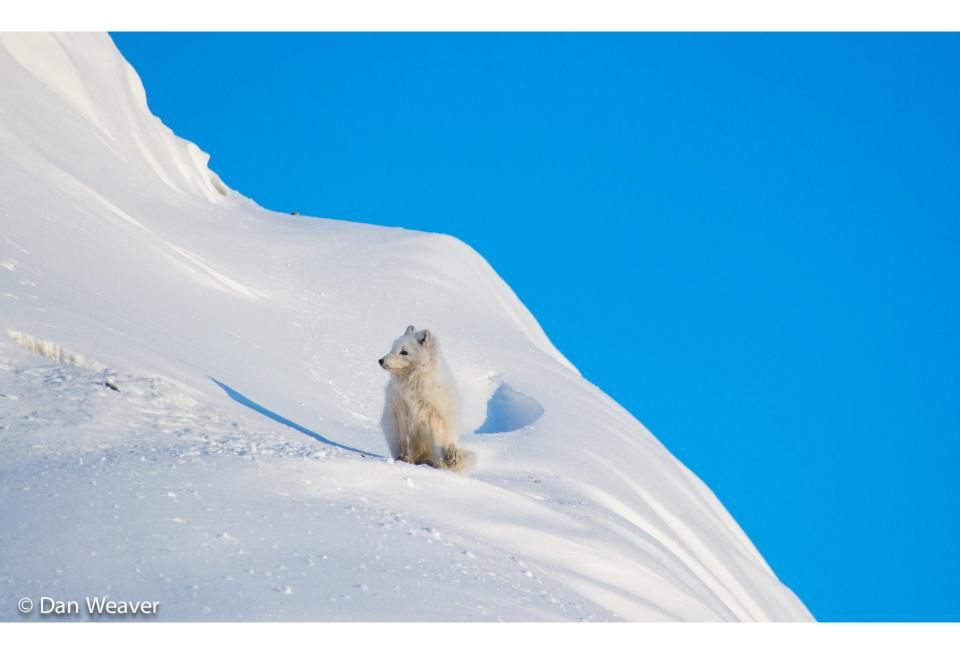






























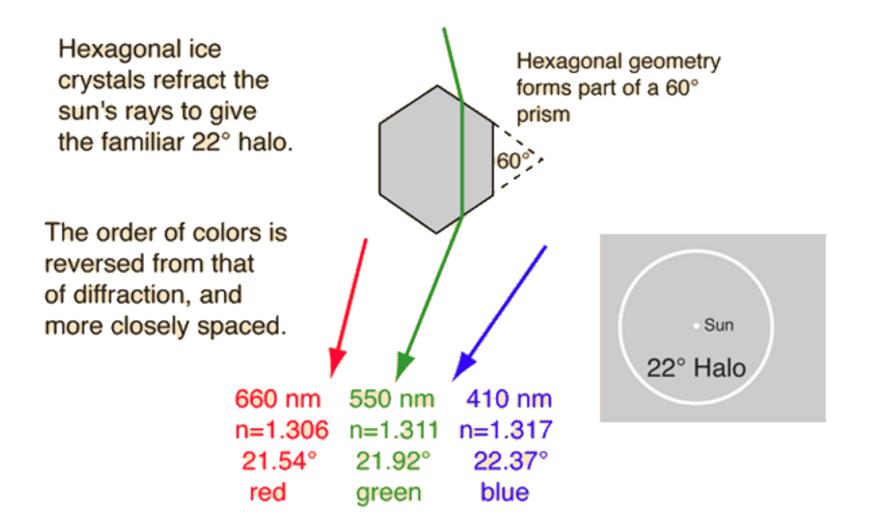




Parhelia, or "sun dogs" (parhelion is singular)



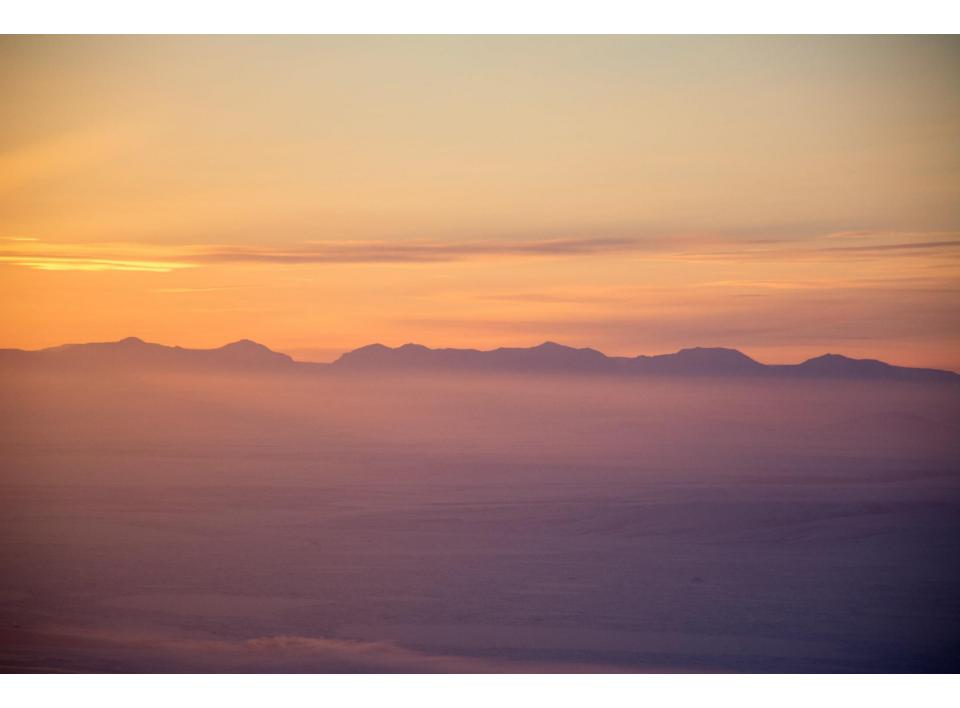
Halo & sundog explanation



Source: http://hyperphysics.phy-astr.gsu.edu/hbase/atmos/halo22.html











Polar Bears

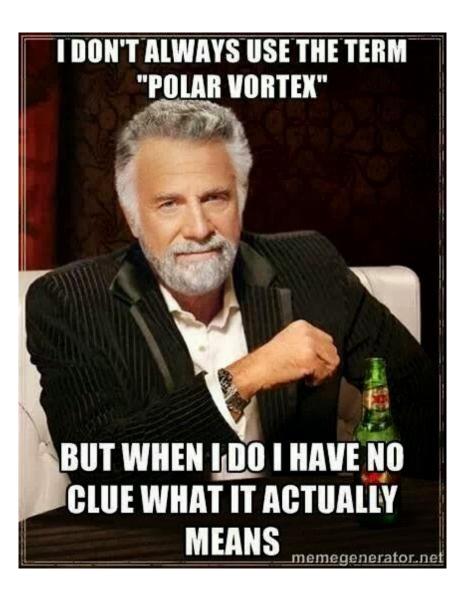
(all photos from Jay Roberts)

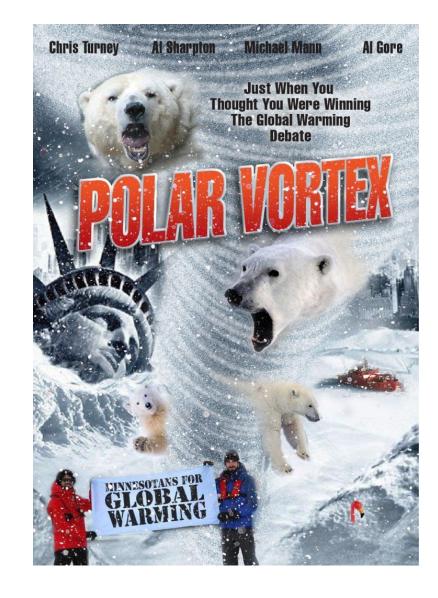


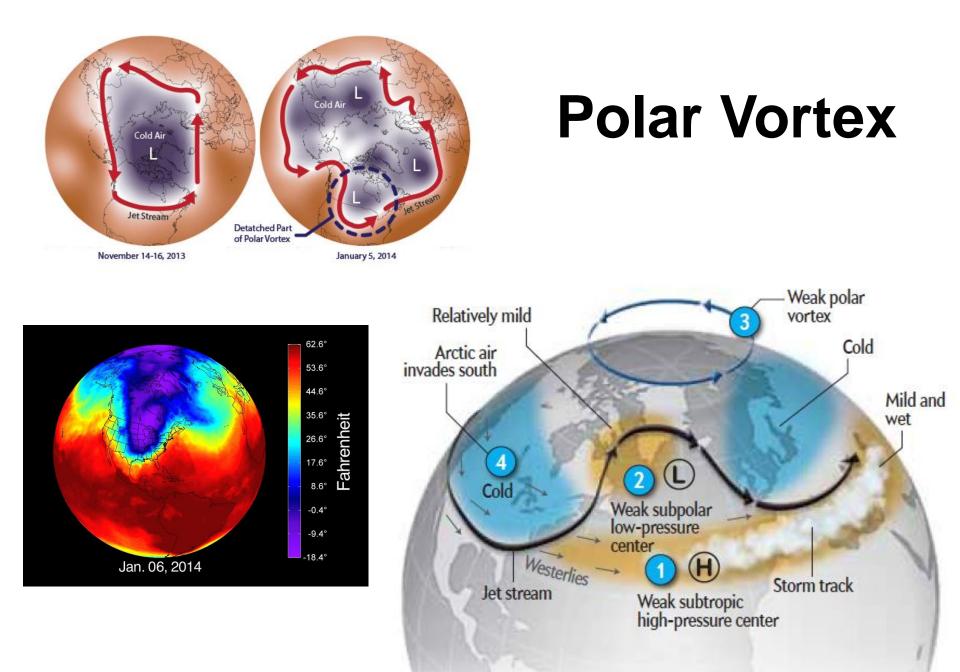


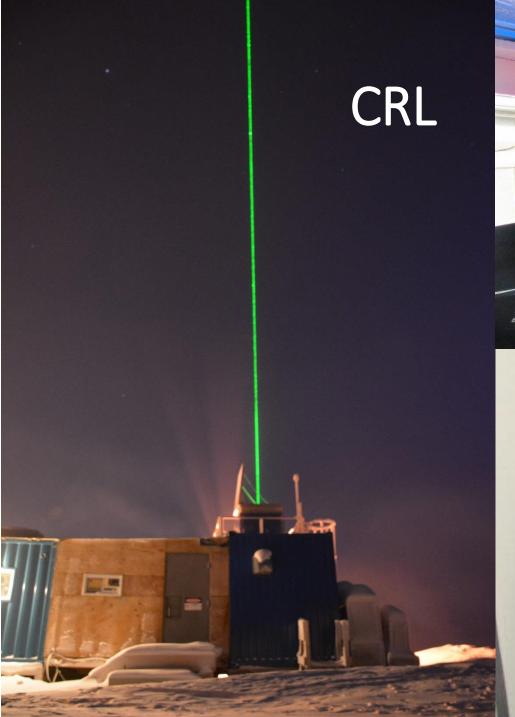


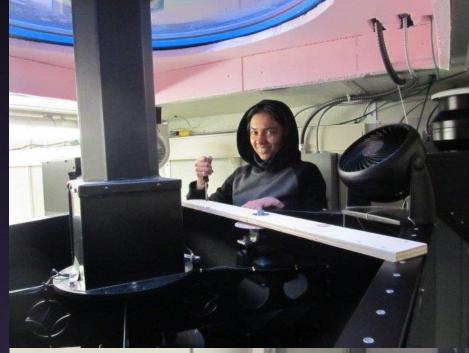
2014 Polar Vortex visits the south

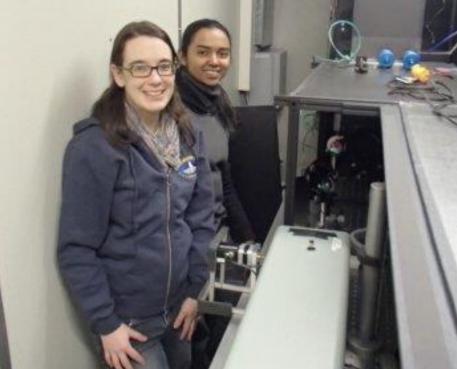












Upper Paradise





Line shapes

- Intrinsically, a spectral line should correspond to one unique energy value
- There are three primary causes of line broadening:
 - Natural broadening (molecule's energy uncertainty)
 - $\,\circ\,$ Has a negligible effect on the observed line shapes. .
 - Pressure broadening (caused by molecular collisions)
 - Dominant effect on the wings of spectra lines for molecules below ~20 km in altitude since it is a pressure-dependent phenomenon (Wallace & Hobbs, 2006).
 - \circ Doppler broadening (caused by motion)
 - Dominant effect on line shapes for molecules above ~ 50 km, but is significant throughout the atmosphere, and is stronger in the center of a spectral line.
- This provides the ability to recover information about the distribution of gases in the atmosphere.
- Often, Pressure and Doppler broadening are combined to be described as a Voigt line shape.

MUSICA microwindows

- Cover a range of weak and strong absorption lines to gain sensitivity to the significant range of abundances.
- These microwindows have four interfering species:
 O₃, CO₂, N₂O, CH₄.
 - Retrieval of CO₂ is used as a quality control test, as abundances are well-known.

